

## Indole Derivatives and a Diketopiperazine from *Chromobacterium violaceum*

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**Abstract** : Three indole derivatives (**1–3**) and a diketopiperazine (**4**) were isolated from the ethyl acetate extract of *Chromobacterium violaceum*. Their structures were elucidated based on the analysis of NMR and HR-MS data and by comparing those in the previous literature. The antibacterial activities of the isolated compounds were evaluated against Gram-positive bacteria, including human pathogenic methicillin-resistant *Staphylococcus aureus* (MRSA), *Lactocaseibacillus paracasei* subsp. *paracasei*, and *Brevibacterium epidermidis*. Compound **1** exhibited moderate antibacterial activity against all the three strains with MIC values ranging from 8.58 to 34.3 µg/mL.

**Keywords** : *Chromobacterium violaceum*, Indole derivatives, Violacein, Diketopiperazine, Antibacterial activity

### Introduction

In the present century, with the increasing threat of antimicrobial resistance, there is an urgent need to develop new antimicrobials.<sup>1</sup> The exploration of natural products from microorganisms remains a crucial strategy for addressing antibiotic resistance. Microbe-derived metabolites have played a fundamental role in antibiotic drug development.<sup>2</sup> The diversity of their structures serves as a rich source of inspiration to identify novel scaffolds for the development of new classes of antibiotics with novel mechanisms of action.<sup>3</sup>

*Chromobacterium violaceum* isolated from water and soil in tropical and subtropical regions is a facultatively anaerobic, Gram-negative, non-sporing, motile, catalase-positive, oxidase-positive, and rod-shaped bacteria.<sup>4</sup> They are found to be a sessile bacterium in their natural habitat. As sessile bacteria are more prone to predation, their secondary metabolites could work as natural defense mechanisms, thereby giving them a competitive advantage

in the habitat.<sup>5</sup> Although *C. violaceum* is recognized for its ability to produce potent antibiotics like violacein, much of the research has centered on violacein itself.<sup>6–8</sup> Therefore, with limited exploration into new antibiotics produced by this strain, our study aims to isolate and verify the antibiotic efficacy of these compounds.

### Experimental

**General experimental procedures** – HR-ESI-MS data was obtained on an Agilent 6545 LC/QTOF spectrometer coupled to an Agilent Infinity II 1290 high-performance liquid chromatography (Agilent Technologies, Santa Clara, CA, USA) with a ZORBAX RRHD Eclipse Plus C<sub>18</sub>, 95 Å (2.1 mm × 50 mm i.d., 1.8 µm, Agilent Technologies, Santa Clara, CA, USA). <sup>1</sup>H-NMR spectra were recorded on a Bruker AVANCE III 700 NMR spectrometer (Varian UNITY INOVA 500 spectrometer, USA). HPLC was conducted using an Agilent Infinity II 1260 High-performance liquid chromatography (HPLC) (Agilent Technologies, Santa Clara, CA, USA) with a column of Luna C<sub>18</sub>(2) 100 Å (250 mm × 10.0 mm i.d., 10 µm, Phenomenex, Torrance, CA, USA). Incubator IB3-03A (Jeiotech, Korea) and Shaking Incubator ISS-3075R (Jeiotech, Korea) were used for incubation of bacteria.

**Bacterial sources** – *C. violaceum* KCTC 2897 (or

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ATCC 12472), *Brevibacterium epidermidis* KCTC 3090, and *Lacticaseibacillus paracasei* subsp. *paracasei* KCTC 2897 were purchased from the Korean Collection for Type Cultures (KCTC), in August 2020. The frozen stocks of the bacteria were deposited in the  $-80^{\circ}\text{C}$  freezer of the School of Pharmacy, Sungkyunkwan University, Suwon, Korea. Methicillin-resistant *Staphylococcus aureus* (MRSA) USA300 was provided by Prof. Wonsik Lee, School of Pharmacy, Sungkyunkwan University.

**Cultivation and extraction** – After taking out the *C. violaceum* frozen stock stored in the  $-80^{\circ}\text{C}$  freezer, it was streaked on Luria-Bertani (LB) agar using a wood stick and incubated for overnight at  $37^{\circ}\text{C}$ . A single colony was collected from the LB agar plate, inoculated using a wood stick to 5 mL of LB broth in the polystyrene tube, and incubated for overnight in a shaking incubator at  $37^{\circ}\text{C}$  and 250 rpm. *C. violaceum* culture was 1:200 diluted by putting 25  $\mu\text{L}$  of the bacterial culture into six 14-mL polypropylene tubes containing 5 mL of LB broth and incubated for 2 days in a shaking incubator at  $37^{\circ}\text{C}$  and 250 rpm. *C. violaceum* culture in the six tubes were inoculated in each 4 L Erlenmeyer flasks with 1 L of LB broth. These cultures were incubated for 2 days in the shaking incubator at  $37^{\circ}\text{C}$  and 250 rpm, which were extracted with ethyl acetate (6 L  $\times$  2 times).

**Isolation and purification of the compounds** – The dried extract (589 mg) was fractionated by HPLC using a  $\text{C}_{18}(2)$  column with a gradient solvent system ( $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ : 10/90 to 100/0 with 0.01% trifluoroacetic acid over 30 mins; flow rate: 4 mL/min; fraction collection: 1 min time window) to afford 30 fractions. Violacein (**1**) was eluted from fraction 17. It was purified with a  $\text{C}_{18}(2)$  column and eluted at a retention time of 18.1 min (1.4 mg) under gradient solvent conditions ( $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ : 30/70 to 40/60 with 0.01% trifluoroacetic acid over 30 min; flow rate: 4 mL/min). Deoxyviolacein and 1-acetyl- $\beta$ -carboline (**2** and **3**) were eluted from the fraction 20. Deoxyviolacein (**2**) was purified ( $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ : 20/80 to 50/50 with 0.01% trifluoroacetic acid over 30 min; flow rate: 4 mL/min) with an identical column and was obtained in its pure form after 25.5 min (0.8 mg). 1-Acetyl- $\beta$ -carboline (**3**) was purified with a  $\text{C}_{18}(2)$  column and eluted at a retention time of 22.5 min (0.8 mg) under gradient solvent conditions ( $\text{CH}_3\text{CN}-\text{H}_2\text{O}$ : 30/70 to 40/60 with 0.01% trifluoroacetic acid over 30 min; flow rate: 4 mL/min). Cyclo(L-Phe-L-Phe) (**4**) was eluted from the fraction 5. It was also purified ( $\text{CH}_3\text{CN}-$

$\text{H}_2\text{O}$ : 25/75 to 35/65 with 0.01% trifluoroacetic acid over 30 min; flow rate: 4 mL/min) with an identical column and was obtained in its pure form at 15.2 min (0.7 mg).

**Violacein (1)** – Purplish gum; HR-ESI-MS (Positive-ion mode)  $m/z$  344.1032  $[\text{M}+\text{H}]^+$  (calcd for  $\text{C}_{20}\text{H}_{14}\text{N}_3\text{O}_3$ , 344.1030);  $^1\text{H-NMR}$  ( $\text{DMSO}-d_6$ , 700 MHz):  $\delta$  11.88 (1H, s, NH-1), 10.71 (1H, s, NH-10), 10.60 (1H, s, NH-15), 9.32 (1H, s, OH-6), 8.93 (1H, d,  $J = 7.6$  Hz, H-19), 8.07 (1H, s, H-2), 7.55 (1H, s, H-13), 7.34 (1H, d,  $J = 8.6$  Hz, H-8), 7.23 (1H, s, H-5), 7.20 (1H, td,  $J = 7.6, 1.2$  Hz, H-21), 6.95 (1H, td,  $J = 7.6, 1.2$  Hz, H-20), 6.82 (1H, d,  $J = 7.6$  Hz, H-22), 6.79 (1H, dd,  $J = 8.6, 2.3$  Hz, H-7).

**Deoxyviolacein (2)** – Purplish gum; HR-ESI-MS (Positive-ion mode)  $m/z$  328.1070  $[\text{M}+\text{H}]^+$  (calcd for  $\text{C}_{20}\text{H}_{14}\text{N}_3\text{O}_2$ , 328.1081);  $^1\text{H-NMR}$  ( $\text{CD}_3\text{OD}$ , 700 MHz):  $\delta$  8.99 (1H, d,  $J = 7.6$  Hz, H-19), 8.01 (1H, dd,  $J = 6.7, 2.0$  Hz, H-5), 7.94 (1H, s, H-2), 7.71 (1H, s, H-13), 7.51 (1H, dd,  $J = 6.7, 2.0$  Hz, H-8), 7.29 (2H, m, H-6 and H-7), 7.22 (1H, td,  $J = 7.6, 1.2$  Hz, H-21), 6.99 (1H, td,  $J = 7.6, 1.2$  Hz, H-20), 6.87 (1H, d,  $J = 7.6$  Hz, H-22).

**1-Acetyl- $\beta$ -carboline (3)** – Colorless gum; HR-ESI-MS (Positive-ion mode)  $m/z$  211.0860  $[\text{M}+\text{H}]^+$  (calcd for  $\text{C}_{13}\text{H}_{10}\text{N}_2\text{O}$ , 211.0866);  $^1\text{H-NMR}$  ( $\text{CD}_3\text{OD}$ , 700 MHz):  $\delta$  8.47 (1H, d,  $J = 4.9$  Hz, H-3), 8.32 (1H, d,  $J = 4.9$  Hz, H-4), 8.23 (1H, dd,  $J = 7.8, 1.0$  Hz, H-7), 7.71 (1H, d,  $J = 8.2$  Hz, H-10), 7.60 (1H, ddd,  $J = 8.2, 7.0, 1.2$  Hz, H-9), 7.32 (1H, ddd,  $J = 8.0, 7.0, 0.9$  Hz, H-8), 2.83 (3H, s,  $\text{H}_3$ -15).

**Cyclo(L-Phe-L-Phe) (4)** – Colorless gum; HR-ESI-MS (Positive-ion mode)  $m/z$  295.1444  $[\text{M}+\text{H}]^+$  (calcd for  $\text{C}_{18}\text{H}_{18}\text{N}_2\text{O}_2$ , 295.1441);  $^1\text{H-NMR}$  ( $\text{CD}_3\text{OD}$ , 700 MHz):  $\delta$  7.34 (4H, tt,  $J = 7.9, 1.2$  Hz, H-5, H-5', H-7 and H-7'), 7.26 (2H, m, H-6 and H-6'), 7.11 (4H, dd,  $J = 7.9, 1.2$  Hz, H-4, H-4', H-8 and H-8'), 4.09 (2H, dd,  $J = 7.0, 4.1$  Hz, H-1 and H-1'), 2.79 (2H, dd,  $J = 13.7, 4.1$  Hz, H-2a and H-2'a), 2.15 (2H, dd,  $J = 13.7, 7.0$  Hz, H-2b and H-2'b).

**Antibacterial test** – The antibacterial test was performed similarly to previous research.<sup>9–12</sup> The three bacterial strains of MRSA, *L. paracasei* subsp. *paracasei*, and *B. epidermidis* were individually streaked on LB agar and incubated for 2 days at  $37^{\circ}\text{C}$ . A single colony was collected from each LB agar plate and inoculated into 5 mL of LB broth in a 14-mL polypropylene tube. It was incubated overnight in a shaking incubator at  $37^{\circ}\text{C}$ , 250 rpm. Then, the  $\text{OD}_{600}$  value of bacterial culture was adjusted to 0.001. A 196  $\mu\text{L}$  of fresh LB medium and 4  $\mu\text{L}$  of each

compound stock solution (10 mM) dissolved in DMSO were added to the first column of a 96-well plate. Using a multichannel pipet, the solution was serially diluted from the second to the last column of the 96-well plate. Then, 100  $\mu$ L of the bacterial culture mentioned above ( $OD_{600} = 0.001$ ) was added to each well. Each experiment was performed three times. The plate was incubated at 37°C in a standing incubator overnight, and the minimum inhibitory concentration (MIC) value of each compound was acquired from the well where the bacteria were not alive as determined visually.

## Results and Discussion

Three indole derivatives (**1–3**) along with a diketopiperazine (**4**) were isolated from the EtOAc extract of *C. violaceum* culture. The chemical structures of compounds **1–4** were determined as violacein (**1**),<sup>13,14</sup> deoxyviolacein (**2**),<sup>13,14</sup> 1-acetyl- $\beta$ -carboline (**3**),<sup>15</sup> and cyclo(L-Phe-L-Phe) (**4**)<sup>16</sup> (Fig. 1) by comparing their NMR and HR-MS data with those reported in the literatures.

Violacein (**1**) was obtained as a purplish gum. It showed UV-vis absorption peaks at 262, 375, and 573 nm, indicating the presence of a conjugated indole skeleton. Its molecular formula was determined to be  $C_{20}H_{13}N_3O_3$  from the molecular ion peak  $[M+H]^+$  at  $m/z$  344.1032 (calcd for  $C_{20}H_{14}N_3O_3$ , 344.1030) in the positive-ion HR-ESI-MS. The  $^1H$ -NMR spectrum of **1** displayed signals for three NH protons at  $\delta_H$  11.88 (1H, s, NH-1), 10.71 (1H, s, NH-10), and 10.60 (1H, s, NH-15), an OH proton at  $\delta_H$  9.32 (1H, s, OH-6), and nine aromatic protons at  $\delta_H$  8.93 (1H, d,  $J = 7.6$  Hz, H-19), 8.07 (1H, s, H-2), 7.55 (1H, s, H-13), 7.34 (1H, d,  $J = 8.6$  Hz, H-8), 7.23 (1H, s, H-5), 7.20 (1H, td,  $J = 7.6$ , 1.2 Hz, H-21), 6.95 (1H, td,  $J = 7.6$ , 1.2 Hz, H-20), 6.82 (1H, d,  $J = 7.6$  Hz, H-22), and 6.79 (1H, dd,  $J = 8.6$ , 2.3

Hz, H-7). Based on these spectroscopic and spectrometric data analysis and the comparison with literature data, the structure of **1** was determined as violacein.<sup>13,14</sup>

The  $^1H$ -NMR and UV spectra of deoxyviolacein (**2**) closely resembled those of violacein (**1**). The difference of  $^1H$ -NMR in **2** was the presence of peaks for a 1,2-disubstituted benzene ring in **2** [ $\delta_H$  8.01 (1H, dd,  $J = 6.7$ , 2.0 Hz, H-5), 7.51 (1H, dd,  $J = 6.7$ , 2.0 Hz, H-8), and 7.29 (2H, m, H-6 and H-7)] instead of the 1,3,4-trisubstituted benzene ring in **1**. HR-ESI-MS data of **2** indicated the molecular formula of **2** as  $C_{20}H_{13}N_3O_2$ , OH less than **1** and subsequently absence of the hydroxyl group at C-6 in **1**. Collectively, the structure of **2** was characterized as deoxyviolacein by comparing its spectroscopic data with the literature values.<sup>13,14</sup>

1-Acetyl- $\beta$ -carboline (**3**) was isolated as a colorless gum. Its molecular formula was established to be  $C_{13}H_{10}N_2O$  based on positive-ion HR-ESI-MS, detected at  $m/z$  211.0860 (calcd for  $C_{13}H_{10}N_2O$ , 211.0866). The  $^1H$ -NMR spectrum of **3** exhibited characteristic  $\beta$ -carboline signals [ $\delta_H$  8.47 (1H, d,  $J = 4.9$  Hz, H-3), 8.32 (1H, d,  $J = 4.9$  Hz, H-4), 8.23 (1H, dd,  $J = 7.8$ , 1.0 Hz, H-7), 7.71 (1H, d,  $J = 8.2$  Hz, H-10), 7.60 (1H, ddd,  $J = 8.2$ , 7.0, 1.2 Hz, H-9), and 7.32 (1H, ddd,  $J = 8.0$ , 7.0, 0.9 Hz, H-8)] along with a single methyl signal at  $\delta_H$  2.83 (3H, s, H<sub>3</sub>-15) resulting from the substituted acetyl group at C-1. The chemical structure of compound **3** was determined by comparing  $^1H$ -NMR and HR-ESI-MS spectral data with those in the literature to be 1-acetyl- $\beta$ -carboline.<sup>15</sup>

Cyclo(L-Phe-L-Phe) (**4**) was purified as a colorless gum. Its molecular formula was confirmed to be  $C_{18}H_{18}N_2O_2$  from the molecular ion peak  $[M+H]^+$  at  $m/z$  295.1444 (calcd for  $C_{18}H_{18}N_2O_2$ , 295.1444) in the positive-ion HR-ESI-MS. Compound **4** was identified through analysis of its  $^1H$ -NMR signals, including symmetric methine protons  $\delta_H$

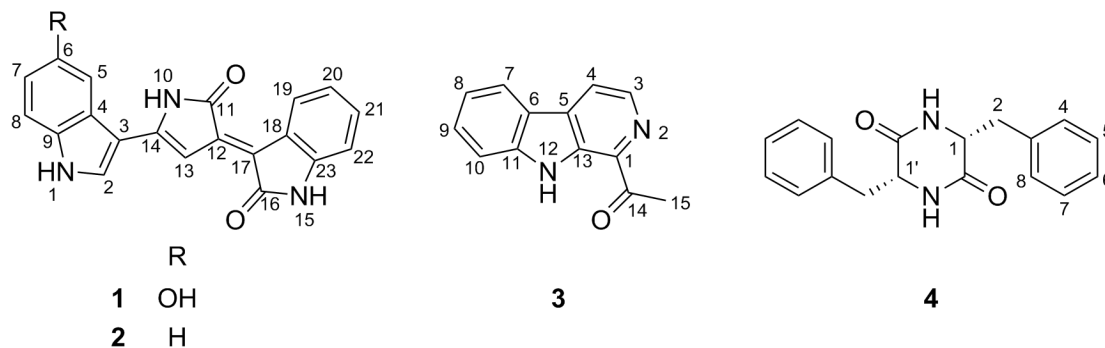


Fig. 1. The structures of **1–4** isolated from *C. violaceum* culture broth.

4.09 (2H, dd,  $J = 7.0, 4.1$  Hz, H-1 and H-1') and methylene protons  $\delta_{\text{H}}$  2.79 (2H, dd,  $J = 13.7, 4.1$  Hz, H-2a and H-2'a) and  $\delta_{\text{H}}$  2.15 (2H, dd,  $J = 13.7, 7.0$  Hz, H-2b and H-2'b), along with signals from monosubstituted benzene protons [ $\delta_{\text{H}}$  7.34 (4H, tt,  $J = 7.9, 1.2$  Hz, H-5, H-5', H-7 and H-7'), 7.26 (2H, m, H-6 and H-6'), and 7.11 (4H, dd,  $J = 7.9, 1.2$  Hz, H-4, H-4', H-8 and H-8')]. The chemical structure of compound **4** was established as cyclic diphenylalanine by analyzing its  $^1\text{H-NMR}$  and HR-ESI-MS spectral data and comparing them with published data.<sup>16</sup>

All the isolated compounds (**1–4**) were evaluated for their antibacterial activities against three Gram-positive bacteria including human pathogenic MRSA,<sup>17–20</sup> *L. paracasei* subsp. *paracasei*, and *B. epidermidis*. Compound **1** exhibited moderate antibacterial activities against all three bacteria with MIC values of 8.6  $\mu\text{g/mL}$  (= 25  $\mu\text{M}$ ) for MRSA, 34.3  $\mu\text{g/mL}$  (= 100  $\mu\text{M}$ ) for *L. paracasei* subsp. *paracasei*, and 17.2  $\mu\text{g/mL}$  (= 50  $\mu\text{M}$ ) for *B. epidermidis* (Table 1). Compounds **2–4** were inactive against the tested bacteria (MIC > 100  $\mu\text{M}$ ). From these antibacterial activity data, we observed an interesting structure-activity relationship (SAR). The only structural difference between **1** and **2** is presence of a hydroxy group at C-6 in **1** which suggested that this functionality would play an important role in displaying antibacterial activity against Gram-positive bacteria. These antibacterial test results of violacein (**1**) are consistent with previous research.<sup>21</sup>

In this study, we report isolation and structure characterization of four secondary metabolites (**1–4**) from *C. violaceum* culture and their antibacterial activity against three Gram-positive bacteria, including a human pathogenic bacterium MRSA. Violacein (**1**) showed moderate antibacterial activity against all three tested bacteria, MRSA, *L. paracasei* subsp. *paracasei*, and *B.*

*epidermidis*, with MIC values of 8.6–34.3  $\mu\text{g/mL}$ . Also, we found that the hydroxy group at C-6 in **1** is essential for its antibacterial activity.

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## Conflicts of Interest

The authors declare that they have no conflicts of interest.

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**Table 1.** Antibacterial activity of compounds from the *C. violaceum*.

Compound	MIC		
	MRSA	<i>L. paracasei</i> subsp. <i>paracasei</i>	<i>B. epidermidis</i>
<b>1</b>	8.6 $\mu\text{g/mL}$ (= 25 $\mu\text{M}$ )	34.3 $\mu\text{g/mL}$ (= 100 $\mu\text{M}$ )	17.2 $\mu\text{g/mL}$ (= 50 $\mu\text{M}$ )
<b>2</b>	> 100 $\mu\text{M}$	> 100 $\mu\text{M}$	> 100 $\mu\text{M}$
<b>3</b>	> 100 $\mu\text{M}$	> 100 $\mu\text{M}$	> 100 $\mu\text{M}$
<b>4</b>	> 100 $\mu\text{M}$	> 100 $\mu\text{M}$	> 100 $\mu\text{M}$

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